## metal-organic papers

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# Peter D.W. Boyd,\* J.B. Challis and C.E.F. Rickard

Department of Chemistry, University of Auckland, Private Bag 92019, Auckland, New Zealand

Correspondence e-mail: pdw.boyd@auckland.ac.nz

#### **Key indicators**

Single-crystal X-ray study T = 150 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.037 wR factor = 0.104 Data-to-parameter ratio = 13.7

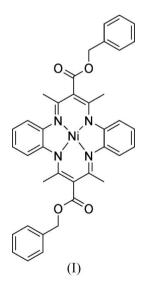
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

## [7,16-Bis(benzyloxycarbonyl)-6,18,15,17tetramethyldibenzo[b,i][1,4,8,11]tetraazacyclotetradecinato- $\kappa^4 N$ ]nickel(II)

The title compound,  $[Ni(C_{38}H_{34}N_4O_4)]$ , is a four-coordinate nickel(II) complex of the 5,15-dihydro-7,16-bis(benzyloxy-carbonyl)-6,18,15,17-tetramethyldibenzo[b,i][1,4,8,11]tetra-azacyclotetradecine ligand, which forms a saddle-shaped molecule with divergent concave surfaces. The molecules assemble in the crystal structure *via* C-H···O interactions with the ester carbonyl O atoms and C-H··· $\pi$  interactions with the propanediiminate portion of the complex.

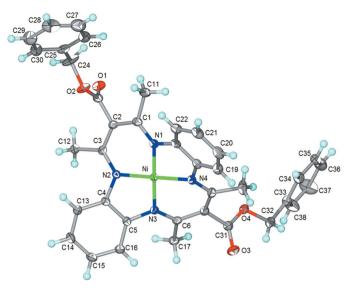
#### Comment

The functionalization of dibenzotetraaza[14]annulene complexes of nickel(II) at the meso positions with alkoxycarbonyl and aryloxycarbonyl groups has recently been reported (Eilmes, Ptaszek & Zielińska, 2001). An extensive set of derivatives has been developed which are of interest as supramolecular hosts (Eilmes, Michalski & Zielińska, 2001; Eilmes et al., 2002, 2003: Rodakiewicz-Nowak et al., 2005). The structure of the title compound (Eilmes et al. 2003), (I), was determined as part of a study of dibenzotetraaza[14]annulene hosts for fullerenes. The divergent concave surfaces of tetraazaannulene complexes have been shown to interact with fullerenes in cocrystallate structures (Andrews et al., 1998, 1999; Croucher et al., 1999)



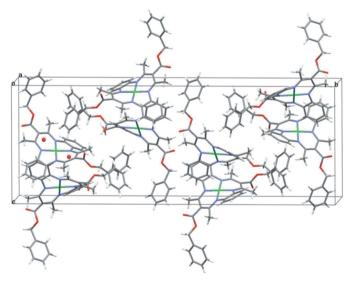
The molecular structure of (I) is shown in Fig. 1. The nickel tetraazaannulene complex is saddle-shaped, with two concave surfaces formed from the NiN<sub>4</sub> plane with either the two benzenoid rings or the two 2,4-pentanediiminate groups. This is due to steric interactions between the methyl groups and the chelate benzene rings (C22-H22···C11 = 2.536 Å, C13-

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The molecular structure of (I), showing 50% probability displacement ellipsoids for non-H atoms and H atoms as arbitrary spheres (X-SEED; Barbour, 2001).





Packing arrangement of complex (I) in the crystal structure, showing the  $C-H\cdots\pi$  interactions to the propanediiminate portion of the complex (MERCURY; Macrae et al., 2006).

H13···C12 = 2.500 Å, C16-H16···C17 = 2.539 Å, C19-H19···C18 = 2.495 Å) (Cotton & Czuchajowska, 1990). The Ni atom, coordinated in a near-planar geometry (Table 1), lies 0.018 (1) Å out of the N<sub>4</sub> plane (r.m.s. deviation 0.0093 Å). The angles between the benzene ring mean planes C9/C10/ C19-C22 (r.m.s. deviation 0.0135 Å) and C4/C5/C13-C16 (r.m.s. deviation 0.0147 Å) and the  $N_4$  plane are 149.50 (9) and  $151.79 (9)^{\circ}$ , respectively, whilst in the other concave surface the angle between the pentanediiminate groups N1/N2 (r.m.s. deviation 0.0191 Å) and N3/N4 (r.m.s. deviation 0.0210 Å) and the N<sub>4</sub> plane are 155.67 (9) and 154.92 (11) $^{\circ}$ , respectively. The angle between the benzene planes is  $121.29 (9)^{\circ}$  and that between the pentanediiminate groups is 130.60 (9)°. These

values are comparable with those observed in the related nickel tetramethyltetraazaannulene complexes (Andrews et al., 1999; Wang et al., 1982). The ester groups on the methine atoms C2 and C7 are arranged anti with respect to the N<sub>4</sub> mean plane.

The molecules of (I) assemble in the crystal structure as shown in Fig. 2. A phenyl ring from the benzyl ester, C25-C30, approaches the centre of a neighbouring complex, with the H atoms on C28 and C29 centred over the N1/N2 and N3/N4 pentanediiminate groups (C28-H28···N3/N4 centroid Cg1 distance = 2.871 Å, C29-H29···N1/N2 centroid Cg2 distance = 2.880 Å). On the opposite face of the complex there are two longer C-H to centroid interactions from C14-H14 in a benzenoid ring of the complex (C14-H14···Cg1 = 2.908 Å) and a methyl group (C12-H12 $C \cdot \cdot \cdot Cg2 = 3.301$  Å). The two ester groups have  $C-H \cdots O$  interactions, one, O2, with an aryl (C16-H16···O3 = 2.427 Å) and methyl (C17-H17A···O3 = 2.392 Å) H atoms, and the other, O1, with aryl (C20-H20...O1 = 2.626 Å and C21-H21...O1 = 2.539 Å)and methyl (C11-H11C···O1 = 2.587 Å) H atoms.

### **Experimental**

Complex (I) was prepared by the method of Eilmes et al. (2003) and characterized by <sup>1</sup>H NMR and FAB mass spectrometry. Crystals were grown by layering of hexane over a CDCl<sub>3</sub> solution of (I).

#### Crystal data

[Ni(C <sub>38</sub> H <sub>34</sub> N <sub>4</sub> O <sub>4</sub> )]	Z = 8
$M_r = 669.40$	$D_x = 1.388 \text{ Mg m}^{-3}$
Orthorhombic, Pbca	Mo $K\alpha$ radiation
a = 10.6079 (2) Å	$\mu = 0.65 \text{ mm}^{-1}$
b = 40.5916 (2) Å	T = 150 (2) K
c = 14.8774 (2) Å	Plate, red
V = 6406.08 (15) Å <sup>3</sup>	$0.36 \times 0.32 \times 0.16 \text{ mm}$

#### Data collection

Siemens SMART CCD areadetector diffractometer (i) scans Absorption correction: multi-scan (SADABS: Sheldrick, 1996)  $T_{\rm min} = 0.799, \ T_{\rm max} = 0.903$ 

#### Refinement

Refinement on  $F^2$  $w = 1/[\sigma^2(F_0^2) + (0.0548P)^2]$  $R[F^2 > 2\sigma(F^2)] = 0.037$ + 14.0271*P*] where  $P = (F_o^2 + 2F_c^2)/3$  $wR(F^2) = 0.104$ S = 0.85 $(\Delta/\sigma)_{\rm max} = 0.011$  $\Delta \rho_{\rm max} = 0.27 \text{ e} \text{ Å}^{-3}$ 5855 reflections  $\Delta \rho_{\rm min} = -0.29 \text{ e } \text{\AA}^{-3}$ 428 parameters H-atom parameters constrained

#### Table 1

Selected geometric parameters (Å, °).

Ni-N1	1.854 (2)	Ni-N4	1.867 (2)
Ni-N3	1.856 (2)	Ni-N2	1.867 (2)
N1-Ni-N3	178.30 (9)	N1-Ni-N2	94.23 (8)
N1-Ni-N4	85.66 (9)	N3-Ni-N2	86.06 (8)
N3-Ni-N4	94.04 (9)	N4-Ni-N2	179.47 (9)

33114 measured reflections

 $R_{\rm int} = 0.035$ 

 $\theta_{\rm max} = 25.4^{\circ}$ 

5855 independent reflections

4985 reflections with  $I > 2\sigma(I)$ 

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H atoms were placed in calculated positions and refined using a riding model [C–H = 0.95–0.99 Å), with  $U_{\rm iso}$ (H) values of 1.2 or 1.5 times  $U_{\rm eq}$ (C). Methyl groups were rotated to fit the H-atom positions to the observed electron density.

Data collection: *SMART* (Siemens, 1995); cell refinement: *SAINT* (Siemens, 1995); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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#### References

Andrews, P. C., Atwood, J. L., Barbour, L. J., Croucher, P. D., Nichols, P. J., Smith, N. O., Skelton, B. W., White, A. H. & Raston, C. L. (1999). J. Chem. Soc. Dalton Trans. pp. 2927–2923.

- Andrews, P. C., Atwood, J. L., Barbour, L. J., Nichols, P. J. & Raston, C. L. (1998). Chem. Eur. J. 36, 2927–2923.
- Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191.
- Cotton, F. A. & Czuchajowska, J. (1990). Polyhedron, 21, 2533-2566.
- Croucher, P. D., Nichols, P. J. & Raston, C. L. (1999). J. Chem. Soc. Dalton Trans. pp. 279–284.
- Eilmes, J., Michalski, O. & Zielińska, K. (2001). Inorg. Chim. Acta, **317**, 103–113.
- Eilmes, J., Ptaszek, M., Dobrzycki, L. & Woźniak, K. (2003). Polyhedron, 22, 3299–3305.
- Eilmes, J., Ptaszek, M. & Woźniak, K. (2002). Polyhedron, 21, 7–17.
- Eilmes, J., Ptaszek, M. & Zielińska, K. (2001). Polyhedron, 20, 143-149.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Macrae, C. F., Edgington, P. R., McCabe, P., Pidcock, E., Shields, G. P., Taylor, R., Towler, M. & van de Streek, J. (2006). J. Appl. Cryst. **39**, 453–457.
- Rodakiewicz-Nowak, J., Nowak, P., Rutkowska-Zbik, D., Ptaszek, M., Michalski, O., Mynarczuk, G. & Eilmes, J. (2005). Supramol. Chem. 17, 643–647.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. Release 97-1. University of Göttingen, Germany.
- Siemens (1995). SMART (Version 4.050) and SAINT (Version 4.050). Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Wang, Y., Peng, S. M., Lee, Y. L., Chuang, M. C., Tang, C. P. & Wang, C. J. (1982). J. Chin. Chem. Soc. 29, 217–224.